

PHAR 632 Review
Oct 9th Workshop

1. One mM of NaCl is equivalent to
 - a. One mOsmolar
 - b. Two mOsmolar
 - c. Half mOsmolar
 - d. Cannot tell as the volume is not identified
2. One mMole of NaCl in 500 mL is equivalent to
 - a. Half mOsmolar
 - b. One mOsmolar
 - c. Two mOsmolar
 - d. Four mOsmolar
3. One molar of dextrose has the same osmolarity as
 - a. Half molar of NaCl assuming complete dissociation
 - b. One molar of NaCl assuming complete dissociation
 - c. Two molar of NaCl assuming complete dissociation
4. The relation between the 2 quantities depends on the total volume.

Rx

Salol 300 mg

Camphor 150 mg

Kaolin 900 mg

M et div. ft. cap. (size 000) No. 3

Which of the following is correct regarding the above prescription?

- a. Each capsule has 300 mg salol
- b. Each capsule has 300 mg Kaolin
- c. Each capsule has 50 mg Camphor
- d. The total capsule content weight is 300 mg
- e. B and C
- f. All of the above

5. Rx

Miconazole 6 g

Corn Starch

Talc aa QS AD 100 grams

Dispense 75 g

If the specific gravity of the powder is 1.2, what size container would you pick?

- a. 1 fl oz
- b. ½ fl oz
- c. 2 fl oz
- d. 3 fl oz

6. Rx
Miconazole 6 g
Corn Starch
Talc aa QSAD 100 grams
Dispense 75 g

What is the percentage strength of miconazole in the final prescription?

- a. 6%
- b. 8%
- c. 10%
- d. 12%

7. Rx
Miconazole 6 g
Corn Starch
Talc aa QSAD 100 grams
Dispense 75 g

How much talc do you need to fill the prescription?

- a. 47 mg
 - b. 35.25 mg
 - c. 100 mg
 - d. 75 mg
 - e. Cannot be calculated
8. What is the ratio strength of 100 mg of drug A is dissolved in 100 mL of hydroalcoholic solution?
- a. 1:1
 - b. 1/10
 - c. 1/100
 - d. 1/1000
 - e. 1/10000

9. What is the concentration of solution containing 0.294g/ml of calcium chloride ($\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$)? (MW=147)
- a. 1mEq/ml
 - b. 2mEq/ml
 - c. 6mEq/ml
 - d. 4mEq/ml

10. The following represents an RBC in hypertonic solution



11. How many mEq of Mg^{2+} ions are there in 5 mL of 20 mg% Magnesium chloride? (Atomic weight Mg =24, Cl=35.5)

- a. 10 mEq
- b. 21 mEq
- c. 33.6 mEq
- d. 50.4 mEq
- e. 420 mEq
- f. 672 mEq
- g. 3200 mEq

12. What is the concentration of KCl in mg/ ml of the solution? Mwt KCl 74.5.

- a. 1490 mg/mL
- b. 1.5 mg/mL
- c. 4470 mg/mL
- d. 15 mg/mL
- e. 0.75 mg/mL

13. The solution in the picture is

- a. Hypertonic
- b. Isotonic
- c. Hypotonic

14. Rx

Miconazole 2 g
 Corn Starch
 Talc aa QS 100 gram
 M ft pulv DTD 200 g

How much talc do you need to fill the prescription?

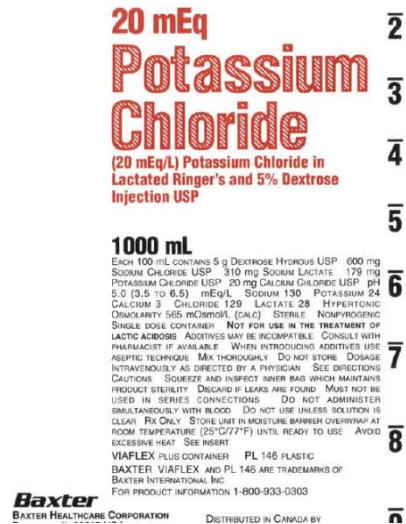
- a. 196 grams
- b. 198 grams
- c. 98 grams
- d. 100 grams

15. The following solutions are isotonic

- a. Normal Saline
- b. Dextrose 5% in water.
- c. A solution with dextrose 2.5% and NaCl 0.45%
- d. A and B
- e. All of the above

16. If both D5W and Normal Saline are isotonic, then one gram of dextrose is equivalent (in terms of isotonic effect) tog of NaCl

- a. 5
- b. 0.18
- c. 0.2
- d. 1
- e. 1.1
- f. 2



17. One mole of NaCl with 90% dissociation will result inmole of particles

- a. 0.9
- b. 1.8
- c. 1.9

18. Rx E

Procaine hydrochloride	1.5%	0.12
Benzalkonium chloride	1:1000	0.13
Boric acid	Qs	0.52
SWFI	QSAD	50 mL

How much Boric acid would you need to fill the prescription?

- | | |
|-----------|----------|
| a. 0.9 g | d. 0.5 g |
| b. 0.35 g | e. 0.2 g |
| c. 0.68 g | |

19. Rx E

Procaine hydrochloride	1.5%	0.12
Benzalkonium chloride	1:1000	0.13
Boric acid	Qs	0.52
SWFI	QSAD	50 mL

How much Benzalkonium chloride would you need to fill the prescription?

- 0.05 g
- 0.1 g
- 20. 0.5 g
- 0.2 g
- 0.01 g

21. Which of these Rx have the same capsule composition?

1- Rx
Salol 300 mg
Camphor 150 mg
Kaolin 900 mg
M et div. ft. cap. (size 000) No. 3

2- Rx
Salol 300 mg
Camphor 150 mg
Kaolin 900 mg
M et ft. cap. (size 000) No. 3

3- Rx
Salol 300 mg
Camphor 150 mg
Kaolin 900 mg
M et ft. DTD cap. (size 000) No. 3

- | | |
|------|---------|
| 1& 2 | 1, 2, 3 |
| 1&3 | none |
| 2&3 | |

22. A pediatric patient is ordered 3 mEq/ kg/ day. If the patient is 17 lb, how many mL of a 23.4% NaCl would be needed?