

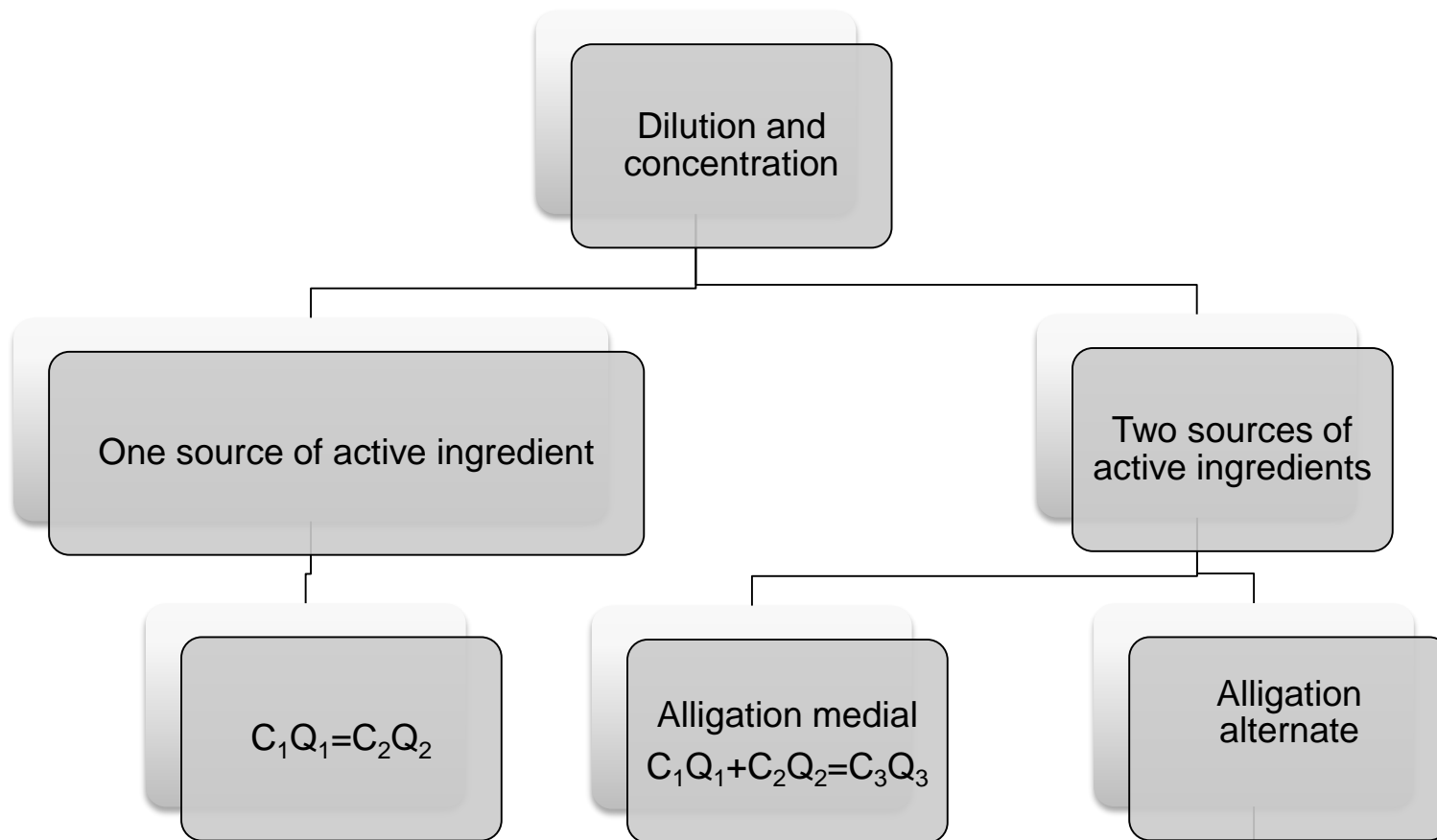
Alligation Dilution and Concentration

Objectives

Students should be able to calculate the necessary quantities needed from different concentrated solutions or preparations in order to prepare a final dosage form with a required concentration.

Why dilution and concentration?

- Preparation of individualized doses (to fulfill the need of different dosage forms &/or strengths from an available product.



High conc preparation		# of parts of the high conc preparation
	Required conc	
Low conc preparation		# of parts of the low conc preparation



Amount of Drug

Quantity of solute = Concentration x quantity of the preparation

How many milliliters of solute are in

1- 6.6 L of 1:3000 v/v solution?

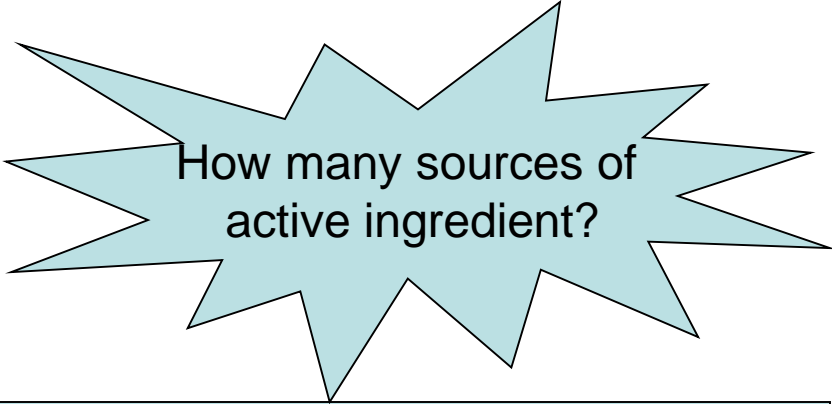
2- 400 mL of 6% v/v solution?



review

Dilution

- Prepare a 30 grams 5% ZnO cream using 20% ZnO cream and white ointment
- CQ equation:



How many sources of active ingredient?

Concentration 1 x quantity 1 = Concentration 2 x quantity 2

Dilution



- A pharmacist needs to prepare 45 grams 10% sulfur in petrolatum using 1: 5 w/w sulfur in petrolatum. How much of each (the medicated and non-medicated creams) does the pharmacist need?

Concentration 1 x quantity 1 = Concentration 2 x quantity 2

Dilution

Remember the 95% is the alcohol concentration

- How much alcohol 95% should be added to 30 mL of 15% w/v solution of drug A, to create a solution of 5% w/v strength of drug A?

Dilution

- A surgeon needs 15 mL of diluted solution of H_2O_2 1:40 v/v. How many mL of water and 20% H_2O_2 should be used?

Combining dosage forms with different concentrations

- Alligation medial (also called Mass Balance Equation),
“thinking in terms of amount of active ingredient”
- Alligation Alternate method
“ thinking in terms of parts”

Alligation medial

Mass Balance Equation

$$C_1Q_1 + C_2Q_2 = C_fQ_f$$

Concentration &
quantity of the final
solution or dosage form

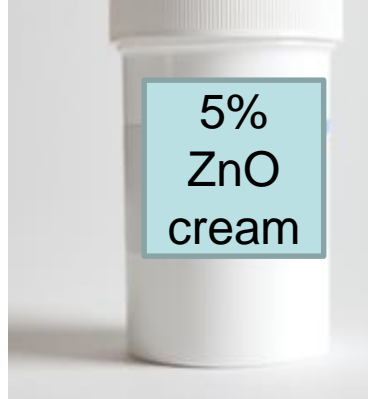
Concentration and quantity of
solution or dosage form 1

Concentration and quantity of
solution or dosage form 2

Remember

$$Q_1 + Q_2 = Q_f$$

Can also be
used if you are
mixing more
than 2 sources
of the active
ingredient



7%



15%



Alligation Medial

Mass Balance Equation

- A pharmacist needs to prepare 28 grams of 10% ZnO cream using 2% , 20% ZnO creams. How much of each cream should be used?

Alligation Medial /

Mass Balance Equation

- How many mL of 15% w/v boric acid and 2.5% w/v boric acid solutions are required to prepare 200 mL of 5% w/v boric acid solution?
- $Q_1 + Q_2 = 200 \text{ mL}$

Alligation Medial / Mass Balance Equation

- How many mL of syrup 80% w/v sucrose should be mixed with 300 mL of syrup 50% w/v sucrose to prepare 70% w/v syrup ?

Alligation Medial / Mass Balance Equation

- Example

Belladonna tincture 20 mL (67% alcohol)

Elixir phenobarbital 70 mL (15% alcohol)

Alcohol USP qs

Simple Syrup qs ad 180 mL

- How much alcohol is needed, so the final solution has 20% alcohol?

Alligation Alternate method

Has multiple uses, but is the method of choice for determine the ratio between 2 components.

- Use 5% and 15 % of drug A creams to prepare 12 % of drug A cream

Strength to be used

15%

Absolute value of the differences

7 part of 15%

Desired strength

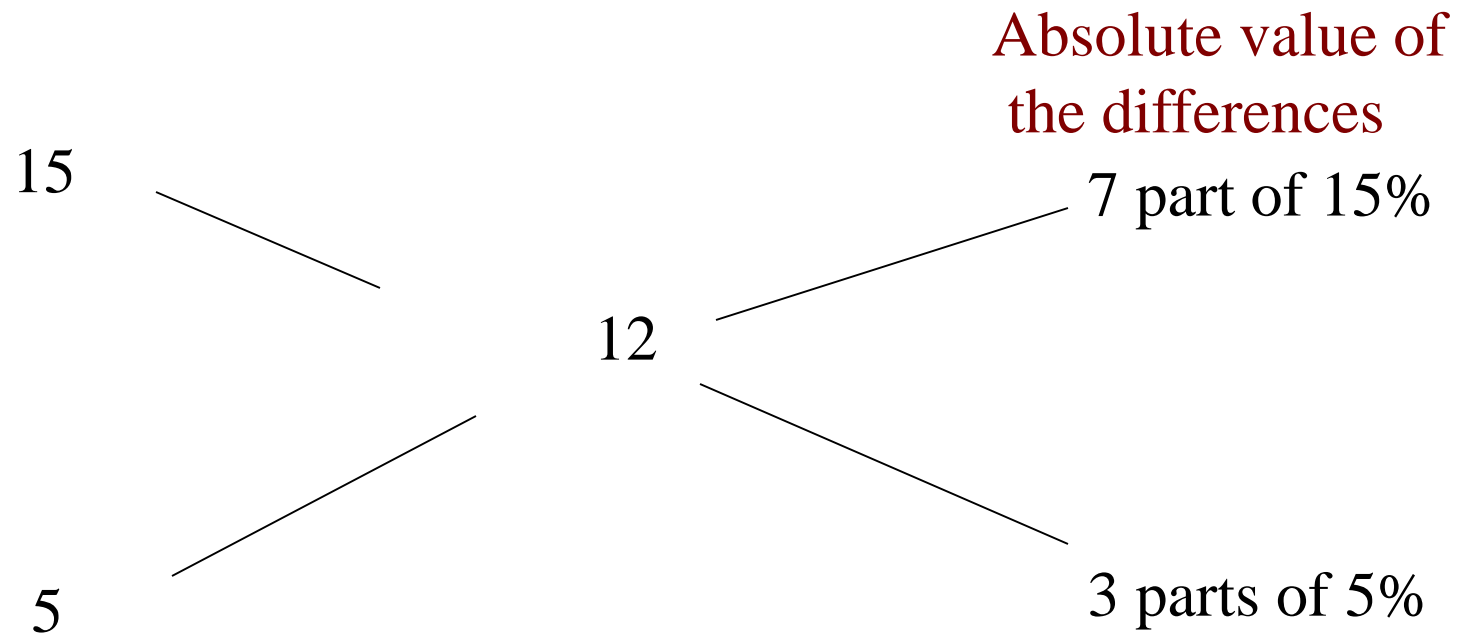
12%

3 parts of 5%

5%

Make sure you have the same unit then think parts

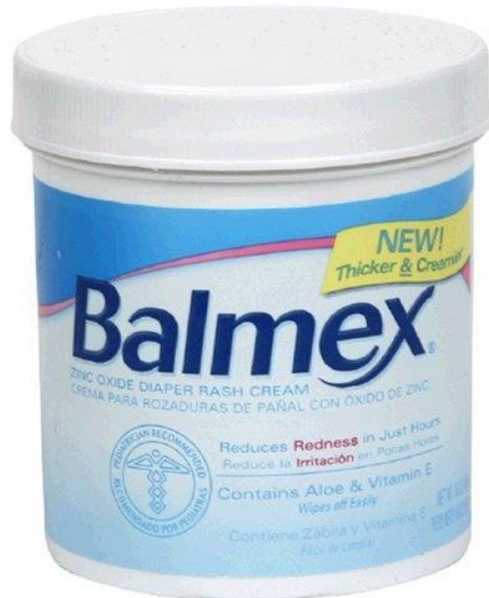
Alligation Alternate method



Application

- How many mL of 2% solution should we mixed with water to prepare 500 mL of solution contains 4 mg/ mL?
- Use Alligation Alternate

- Rx
 - Hydrocortisone 0.1%
 - Zinc oxide 5%
 - Hydrophilic Base qsd 100 gram
 - Dispense 30 g
- How would you prepare this prescription using the following creams?

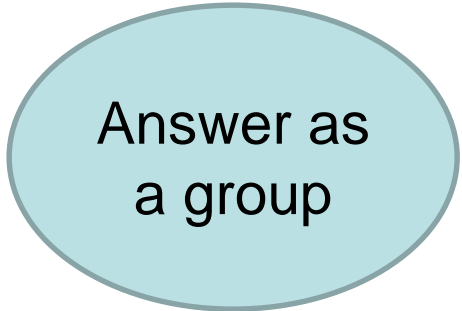


11% ZnO cream



How much of alcohol 50% should be mixed with alcohol 10 % to prepare 40 mL of alcohol 5%?

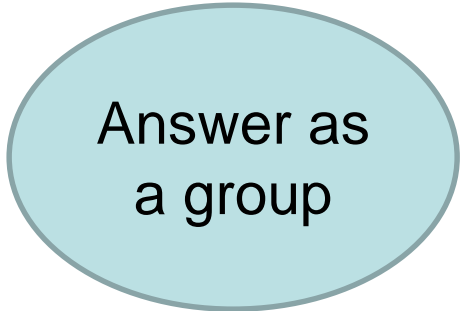
- A. 5 parts of alcohol 50% and 45 parts of alcohol 10%.
- B. 5 parts of alcohol 50% and 40 parts of alcohol 5%.
- C. This concentration cannot be achieved



Answer as
a group

What is the **ratio** of alcohol 90% : alcohol 25% needed to prepare alcohol 60%?

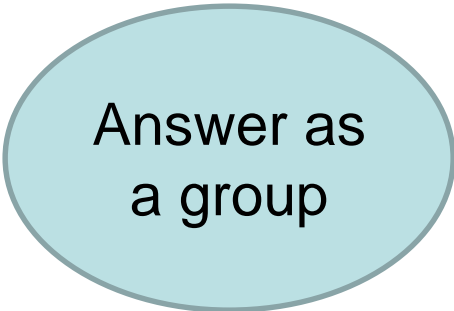
- A. 7:6
- B. 54%
- C. 35:65
- D. 6:7



Answer as
a group

Application

- How many mL of Phenytoin elixir containing 25mg/tsp and 35 mg/tsp should be used to prepare 100 ml of elixir containing 6.2 mg/mL?



Answer as
a group